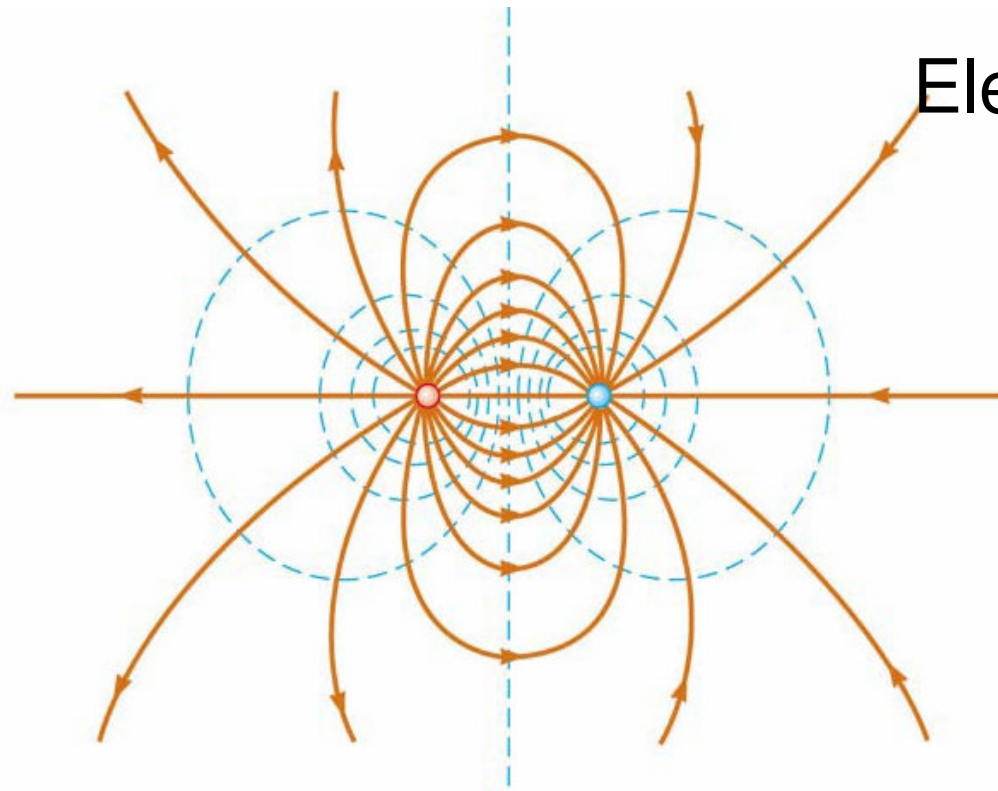
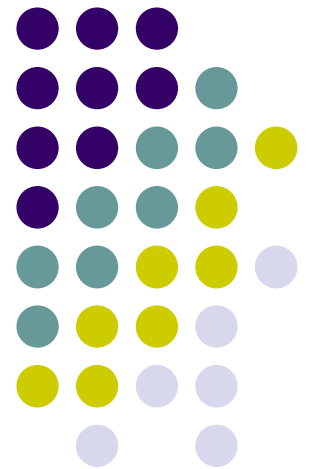


Chapter 23



Electric Potential

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SJSU





Goals for Chapter 23

- To study and calculate **electrical potential energy**
- To define and study examples of **electric potential**
- To trace regions of equal potential as **equipotential surfaces**
- To find the electric field from electrical potential



Electrical Potential Energy

$$F_G = G \frac{m_1 m_2}{r^2} \quad \text{Gravitational force}$$

$$F_e = k \frac{q_1 q_2}{r^2} \quad \text{Coulomb force}$$

- similar to gravitational, electrostatic force is a **conservative** force

$$W_{\text{net}} = \oint \vec{F} \cdot d\vec{\ell} = 0$$

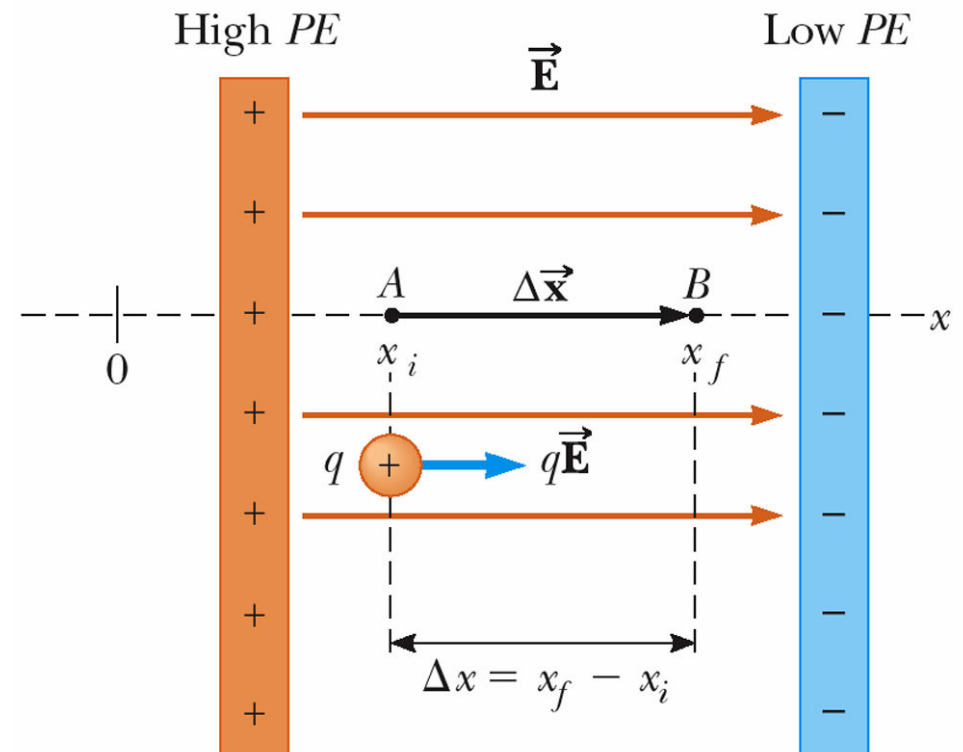
- which means there is a potential energy associated with this force such that

$$F = -\frac{dU}{dx} \quad \text{or gradient in 3D}$$



Work and Potential Energy

- There is a uniform field between the two plates
- As the charge moves from A to B, work is done on it
- $W = Fd = q E_x (x_f - x_i)$
- $\Delta PE = - W$
 - $= - q E_x \Delta x$
 - Only for a uniform field





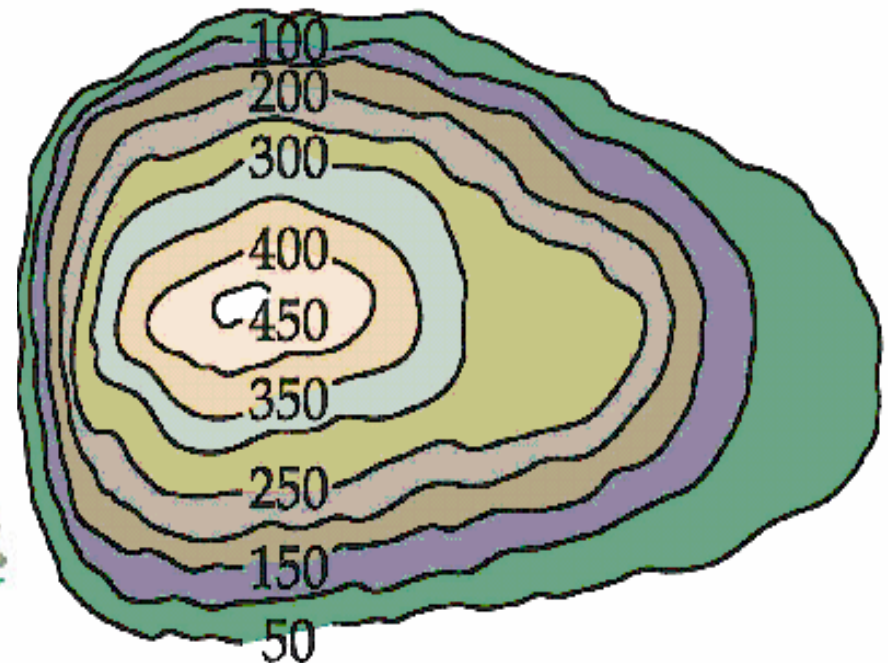
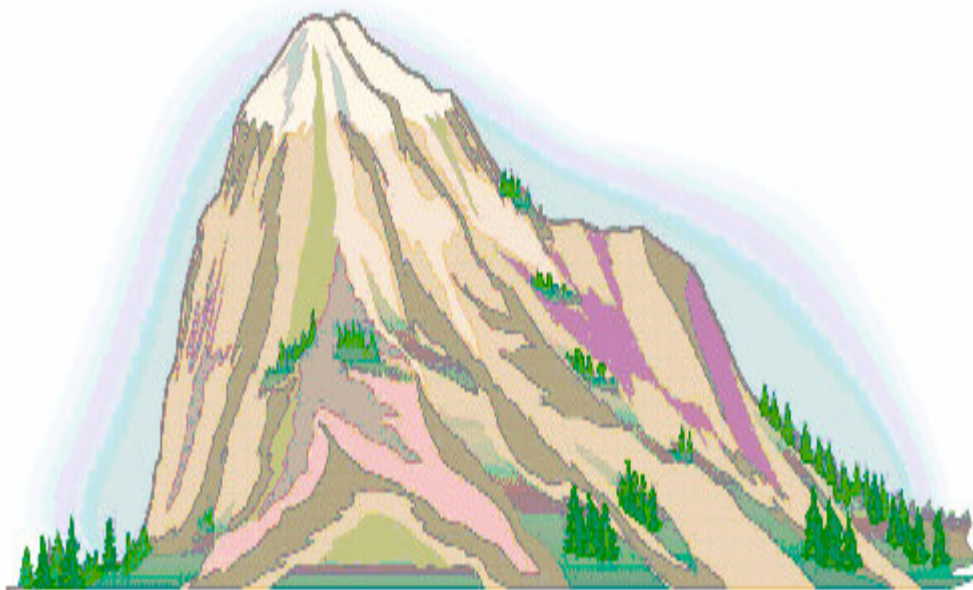
Potential Difference (voltage)

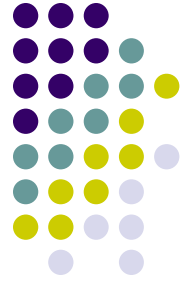
- Potential difference is *not* the same as potential energy
- The potential energy and the potential difference are related by : $\Delta PE = q \Delta V$
- Both electric potential energy and potential difference are *scalar* quantities
- Units of potential difference
 - $V = J/C$
- A special case occurs when there is a *uniform electric field*
 - $\Delta V = V_B - V_A = -E_x \Delta x$
 - Gives more information about units: $N/C = V/m$



Equipotential Contour (2D)

On a contour map, the curves mark constant elevation; the steepest slope is perpendicular to the curves. The closer together the curves, the steeper the slope.





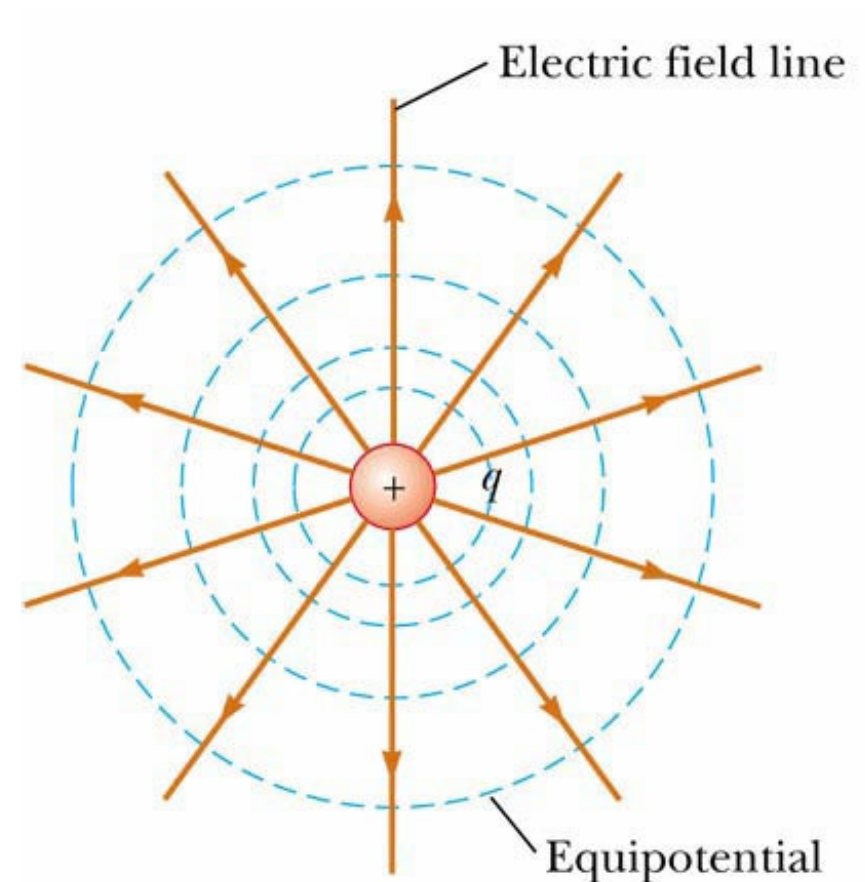
Equipotential Surfaces (3D)

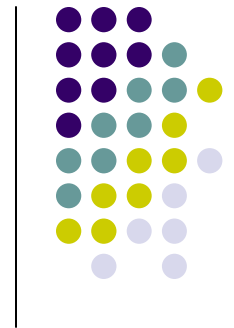
- An *equipotential surface* is a surface on which all points are at the same potential
 - No work is required to move a charge at a constant speed on an equipotential surface
 - The electric field at every point on an equipotential surface is perpendicular to the surface



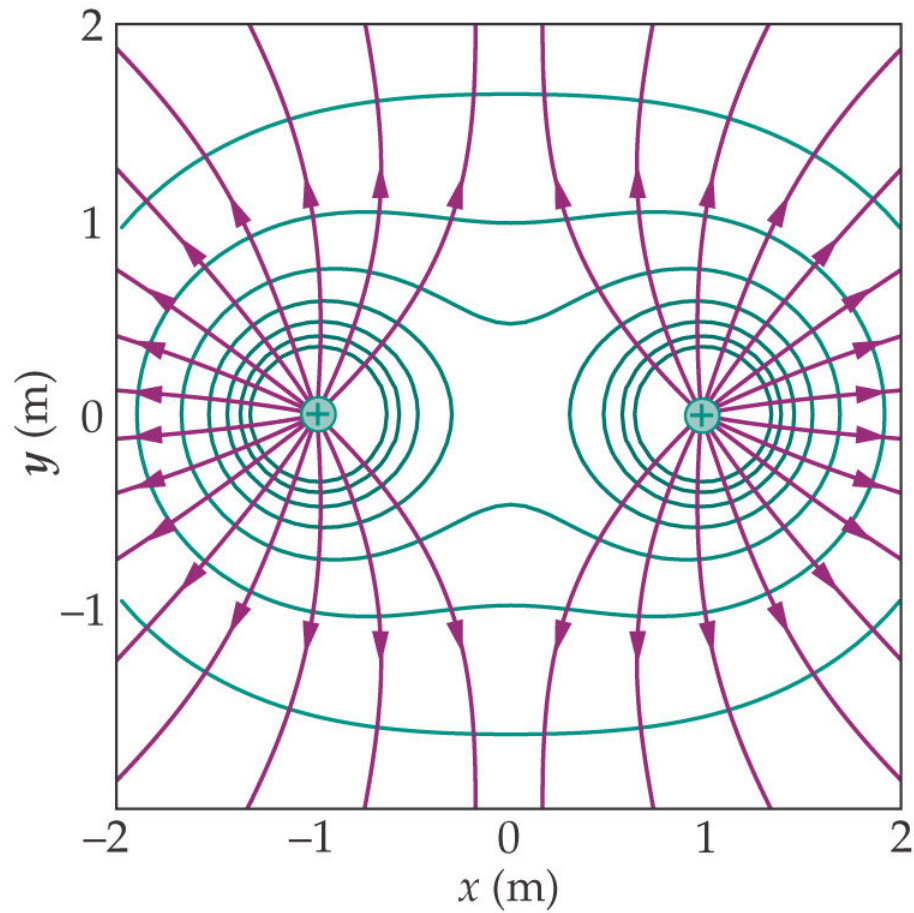
Equipotentials and Electric Fields Lines – Positive Charge

- The equipotentials for a point charge are a family of spheres centered on the point charge
- The field lines are perpendicular to the electric potential at all points

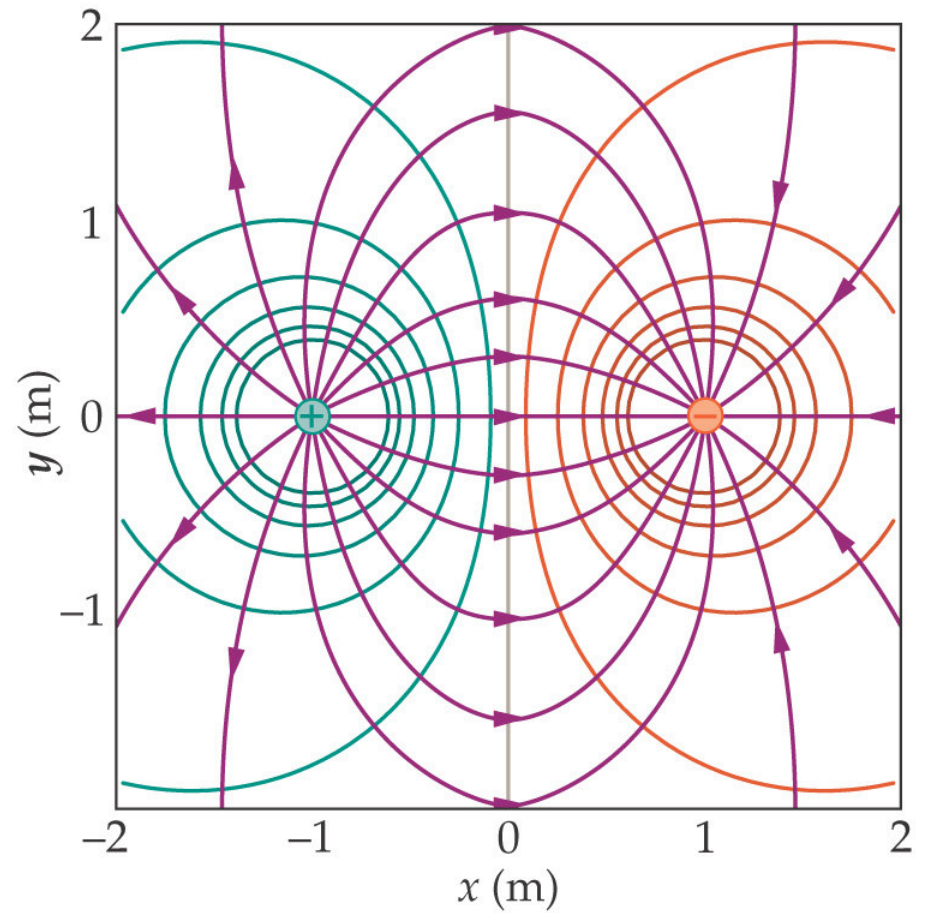




For two point charges



(a)

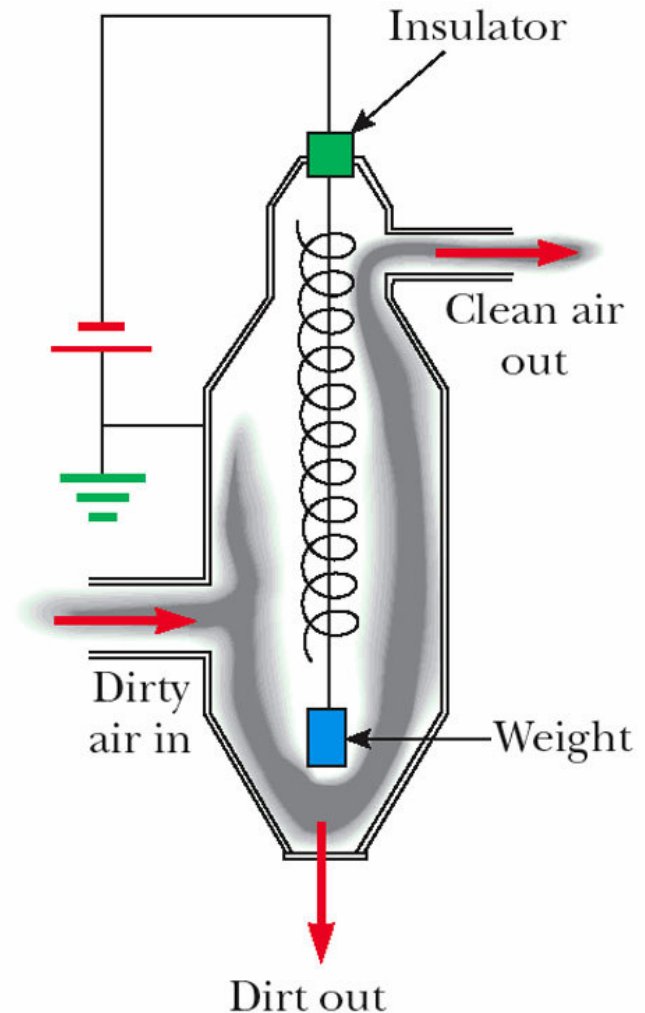


(b)



Application – Electrostatic Precipitator

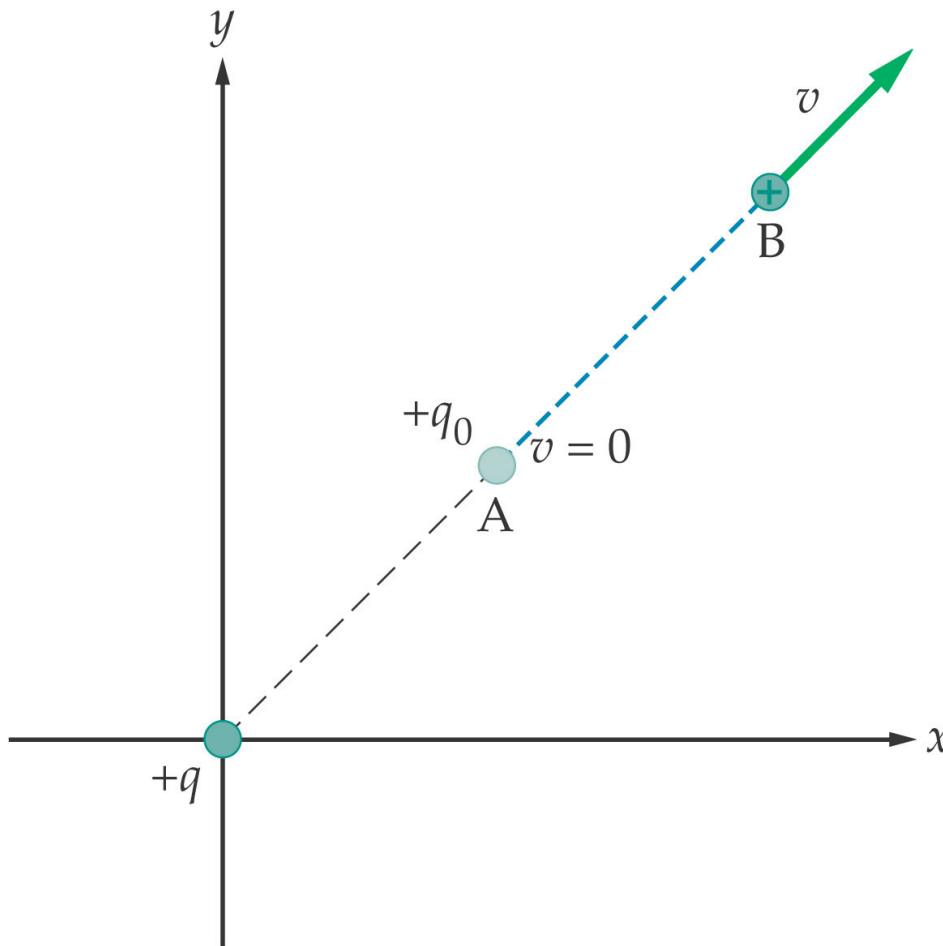
- It is used to remove particulate matter from combustion gases
- Reduces air pollution
- Can eliminate approximately 90% by mass of the ash and dust from smoke
- Recovers metal oxides from the stack





Electric Potential of Point Charges

The difference in potential energy between points A and B is



$$W = \int_{A}^{B} \vec{F} \cdot d\vec{r} = \int_{A}^{B} k \frac{q_0 q}{r^2} dr$$

$$-\Delta U = -k \frac{q_0 q}{r_B} + k \frac{q_0 q}{r_A}$$

$$-(U_B - U_A) = -k \frac{q_0 q}{r_B} + k \frac{q_0 q}{r_A}$$

$$U_A - U_B = k \frac{q_0 q}{r_A} - k \frac{q_0 q}{r_B}$$



The Electric Potential of a Point Charge

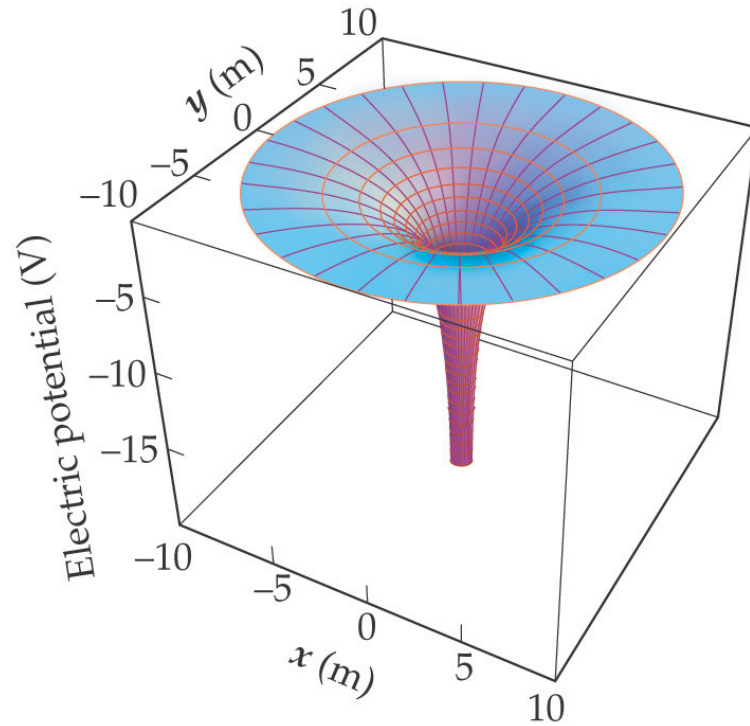
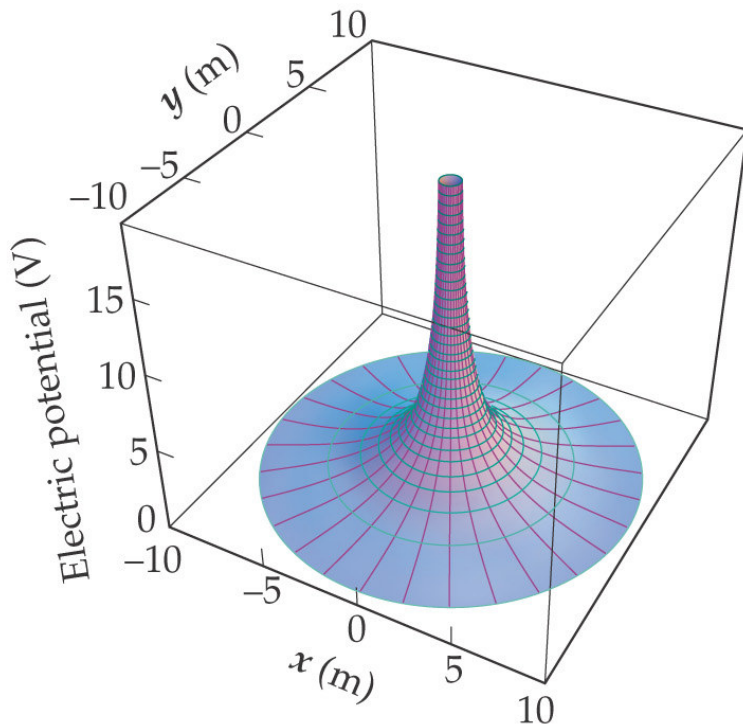


PE between 2 pt charges:

$$U = k \frac{q_1 q_2}{r}$$

Electric potential from 1 pt charge:

$$V = k \frac{q}{r}$$



shown here is V for a positive and negative charge.

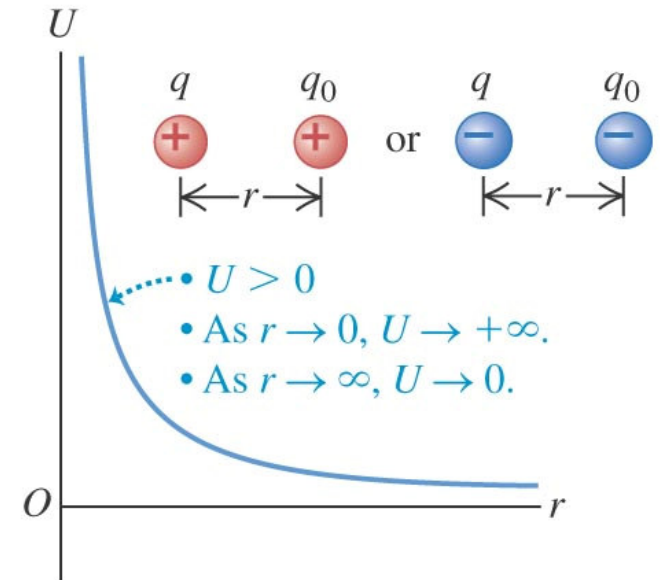
Potential energy curves —PE versus r

$$U = k \frac{q_1 q_2}{r}$$

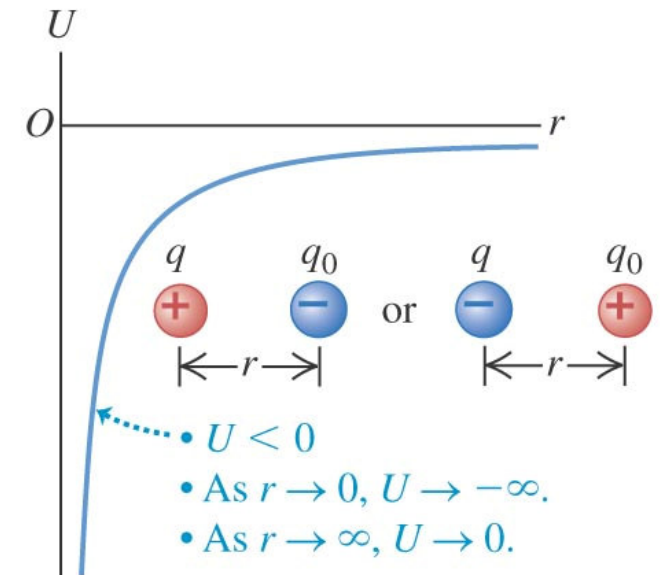
$U > 0$ for like charges.
 $U < 0$ for opposite charges.

$$F = - dU/dr$$

(a) q and q_0 have the same sign.



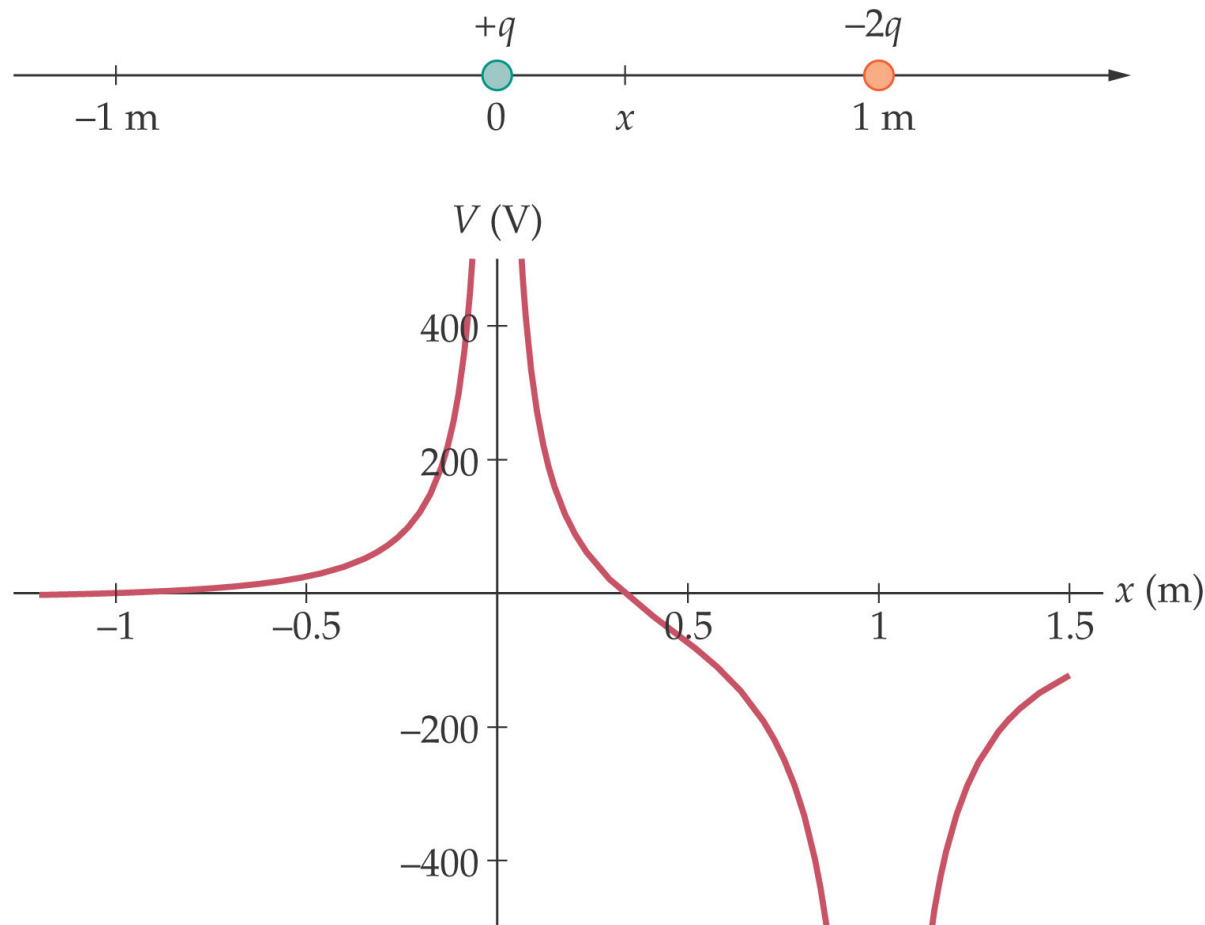
(b) q and q_0 have opposite signs.





The Electric Potential of Point Charges

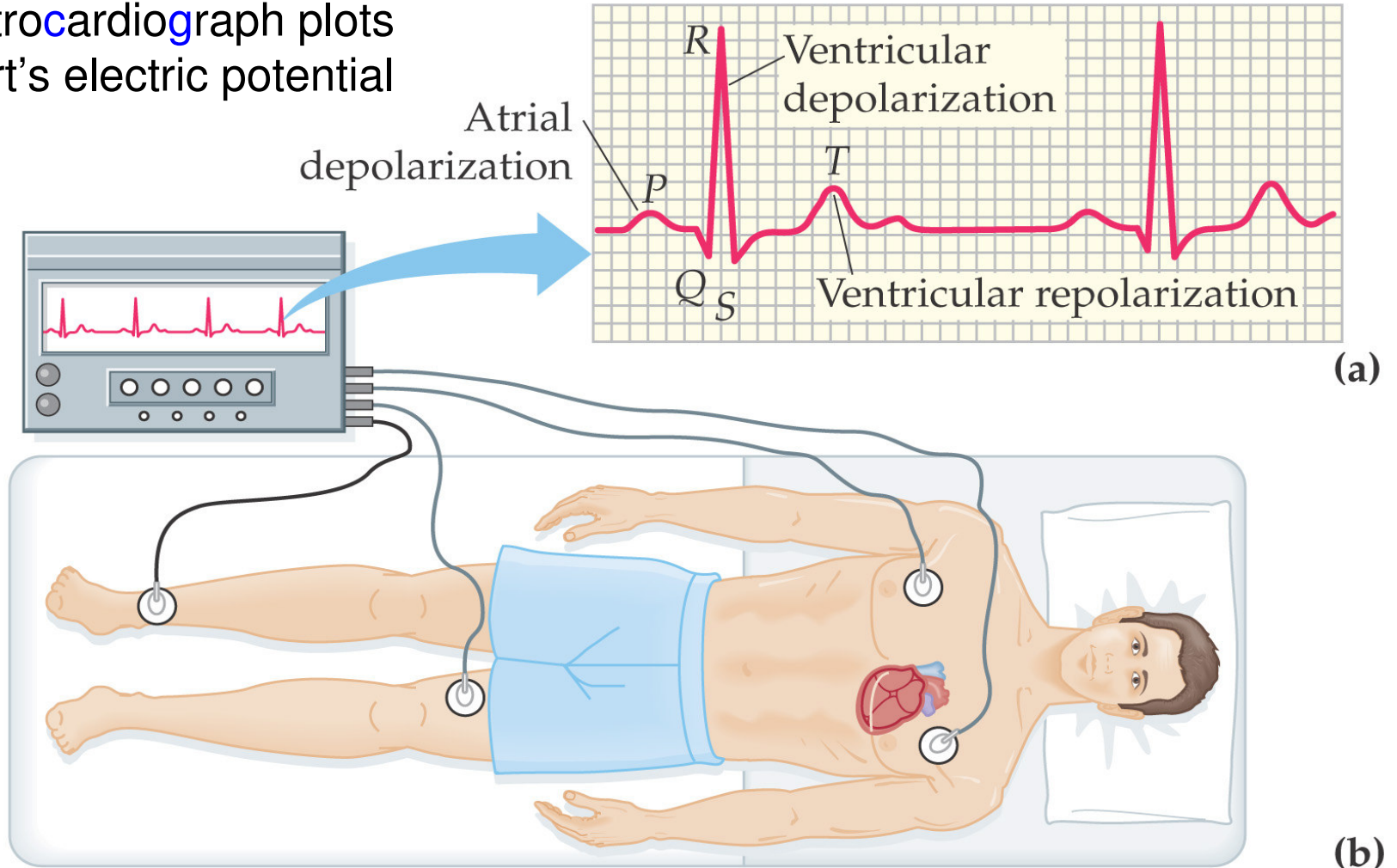
The electric potential of a group of point charges is the algebraic sum of the potentials of each charge.





Human – a complex circuit?

An **electrocardiograph** plots the heart's electric potential
ECG

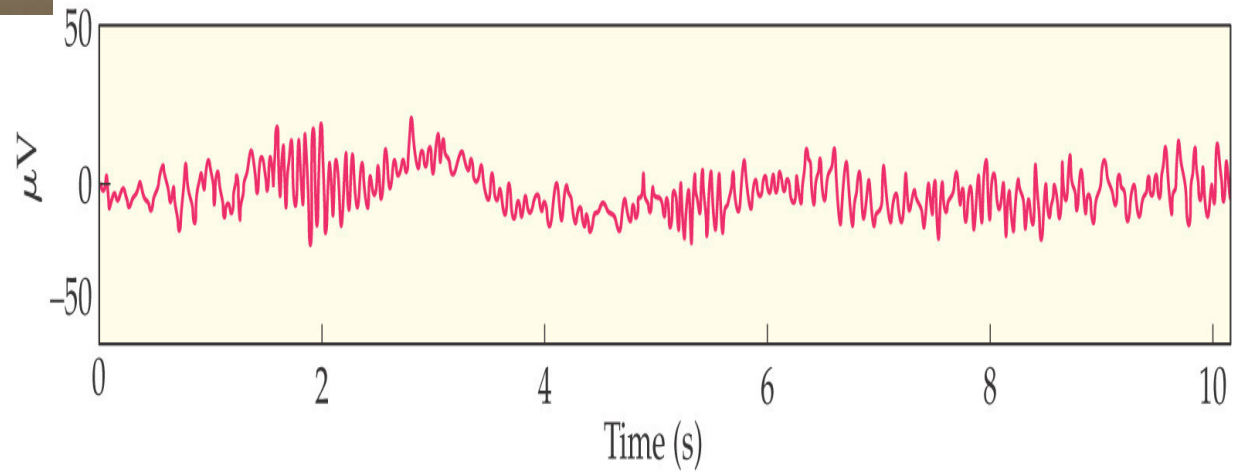




EEG



An **e**lectro**e**ncephalo**g**raph measures the electrical activity of the brain.





The Electron Volt

- The electron volt (eV) is defined as the energy that an electron gains when accelerated through a potential difference of 1 V
 - Electrons in normal atoms have energies of 10's of eV
 - Excited electrons have energies of 1000's of eV
 - High energy gamma rays have energies of millions of eV
- $1 \text{ eV} = 1.6 \times 10^{-19} \text{ J}$

Example

